Objectives

Introduction to EPR theory

- a) The resonance phenomenon
- b) Anisotropy
- c) Saturation
- Detailed protocol on using spin labels to determine protein structure and dynamics
 - a) Nitroxide scanning
 - b) Local environment and secondary structure
 - c) Tertiary and quaternary structure
 - d) Conformational dynamics

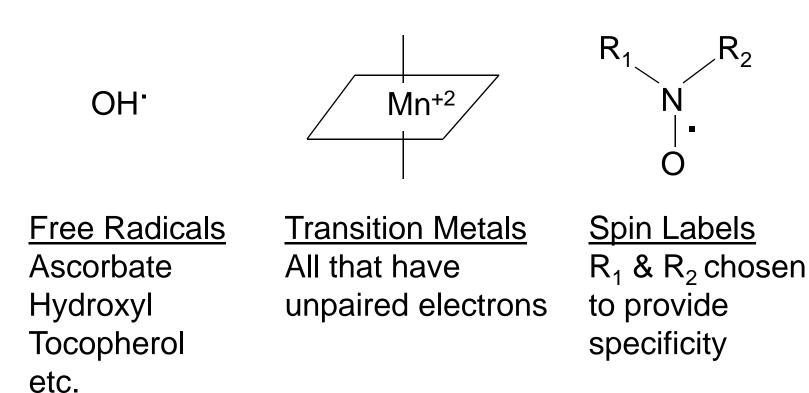
Tools of the Trade

- **1** Atomic resolution: X-ray crystallography
- 2- NMR spectroscopy
- **3** de novo Modeling and structure determination, Homology modeling
- 4- CryoEM

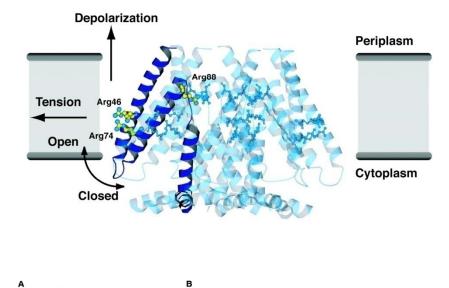
Why do we needs probes ?????

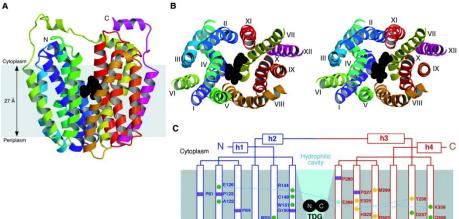
Issues: 1- sample requirement 2- nature of the environment 3- Dynamics and time scales

<u>Electron</u> <u>Paramagnetic</u> <u>Resonance</u> <u>Electron</u> <u>Spin</u> <u>Resonance</u>

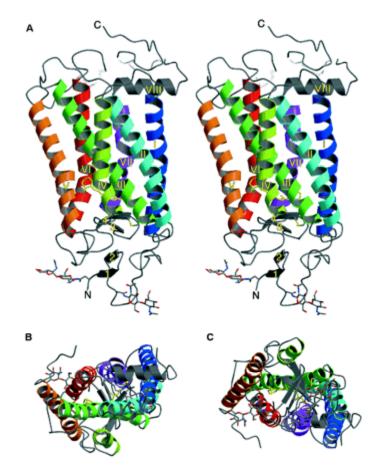


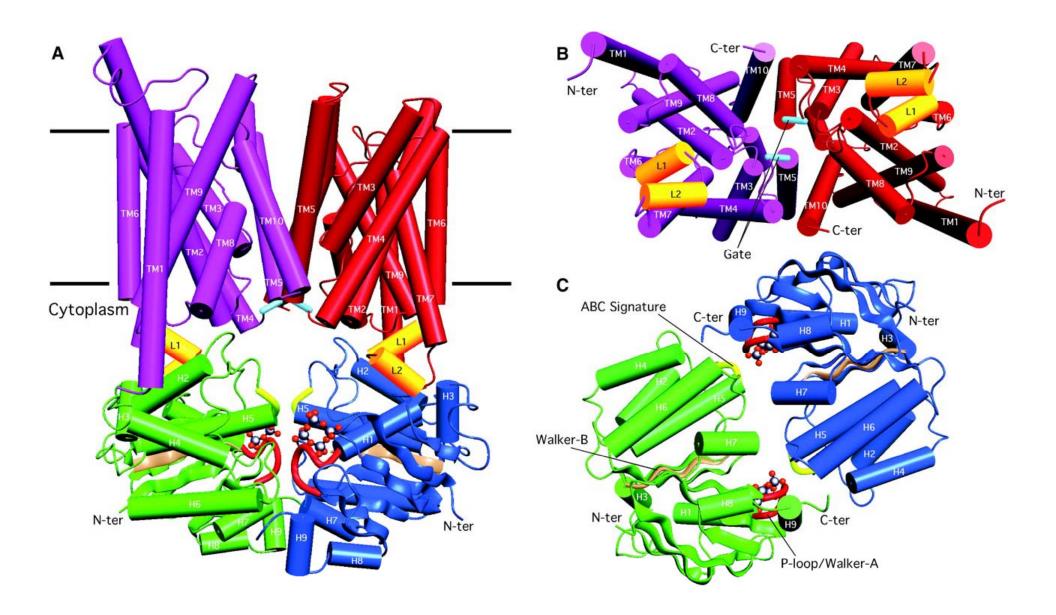
Functional Dynamics of Proteins

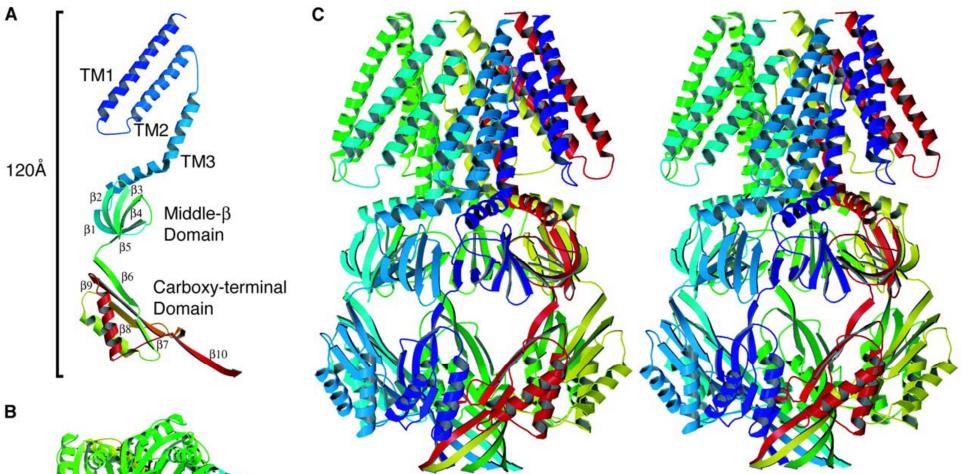


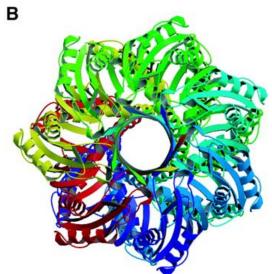


Periplasm





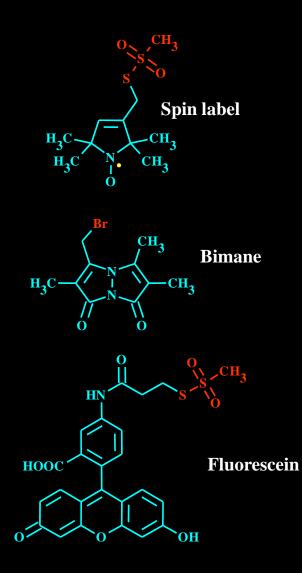




Α

Site-Directed Labeling





Probe spectroscopy

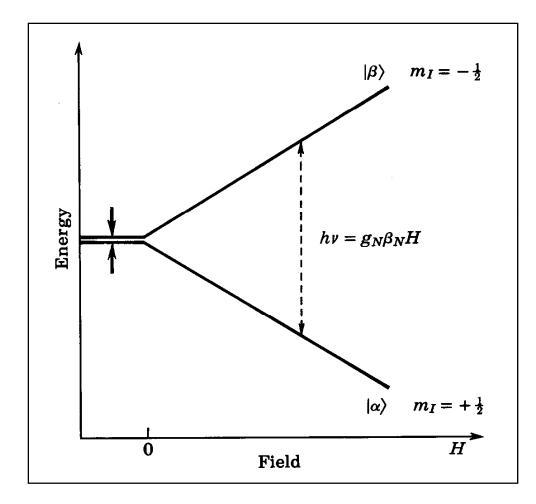
Spectral parameter — Protein structure

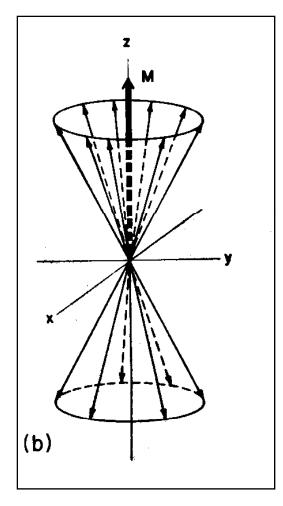
Biggest challenge: What is the transfer function? Examples :

<u>Uncertainty principle of chemical modification:</u> Implications: limited resolution both on the static and dynamic levels

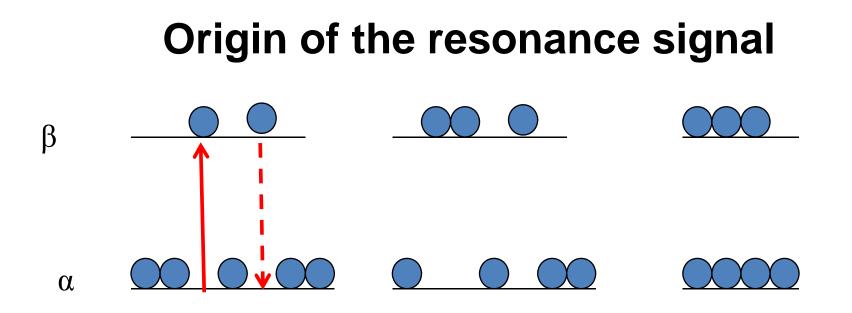
Functional analysis of labeled mutants is a REQUIREMENT

Spin 1/2 Electron in an External Field



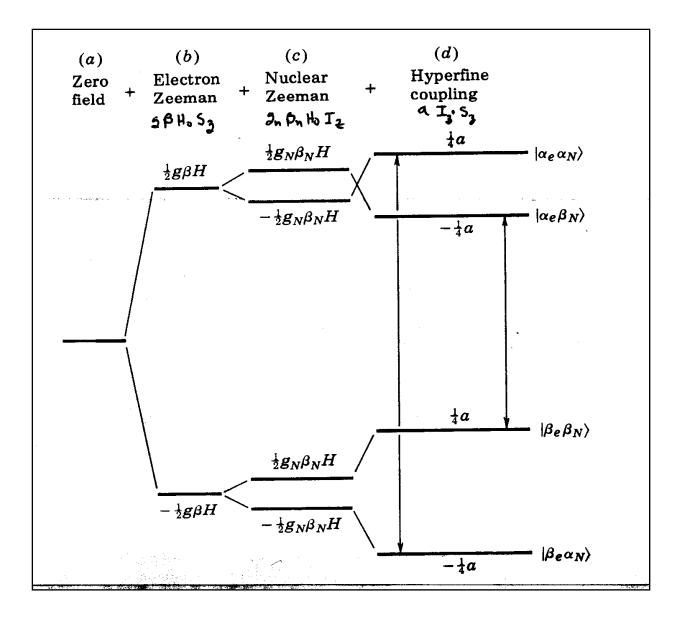


Electron Spins Precess @ the Larmor Frequency about the External Field $\gamma_e = 1.76e^7$ rad/sec/G

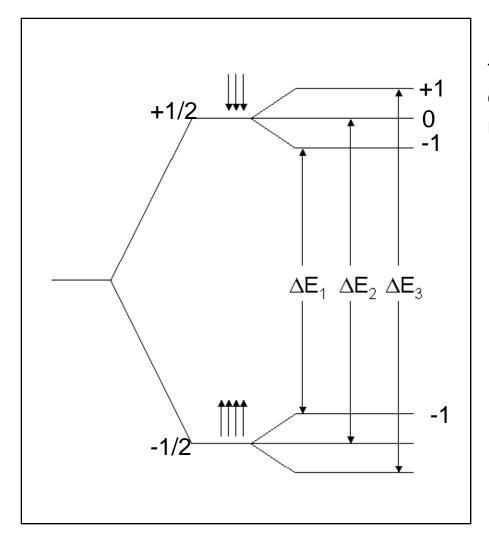


- Difference in population defined by Boltzman Distribution
- •The concept of thermal equilibrium: Relaxation
- •Saturation

Energy level Diagram for a ¹⁵N Nitroxide $S=\frac{1}{2}$; $I=\frac{1}{2}$

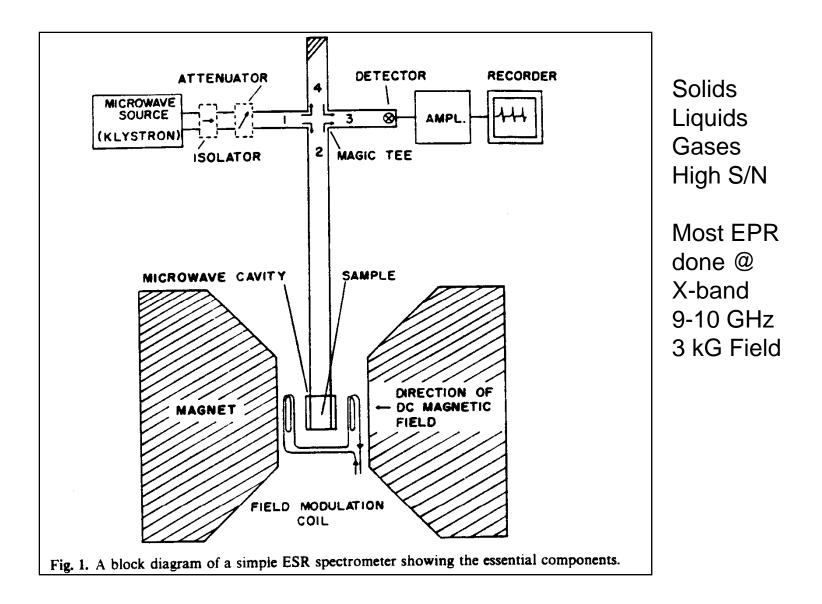


Approximate Energy Levels for a ^{14}N Nitroxide Spin Label; S = $\frac{1}{2}$, I=1

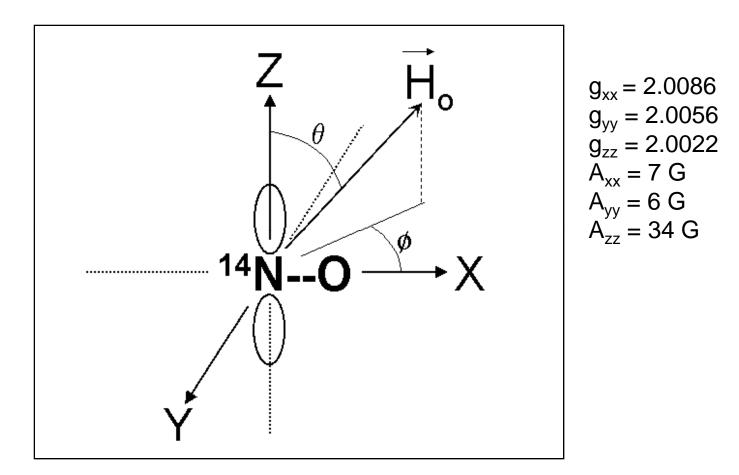


In general, EPR gives information on the # and nuclear spin state of nearby nuclei –coordination sphere of metal centers in metalloproteins

Block Diagram of EPR Spectrometer

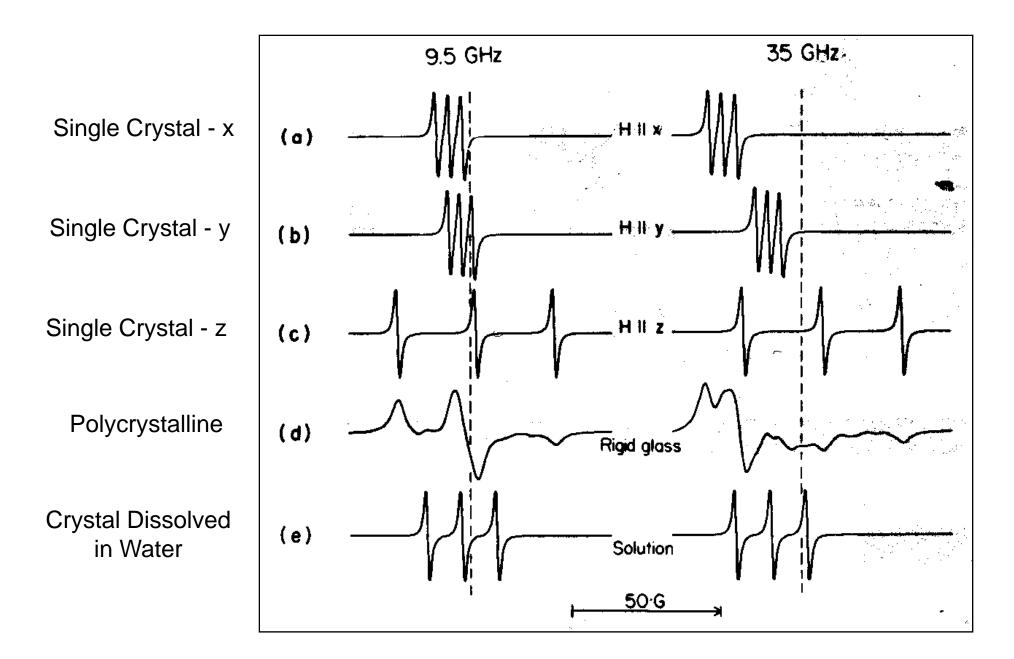


Nitroxide Reference Frame

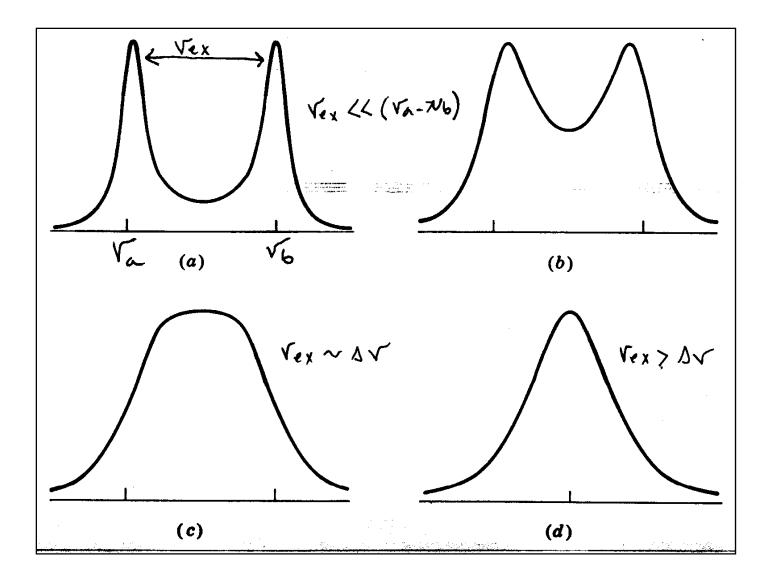


$$\begin{split} H_{res} &= [h\nu/\beta_e g_{eff}(\theta, \phi)] + m_i A_{eff}(\theta, \phi) \\ g_{eff} &= g_{xx} sin^2 \theta cos^2 \phi + g_{yy} sin^2 \theta sin^2 \phi + g_{zz} cos^2 \theta \\ A_{eff} &= [A_{xx}^2 sin^2 \theta cos^2 \phi + A_{yy}^2 sin^2 \theta sin^2 \phi + A_{zz}^2 cos^2 \theta]^{\frac{1}{2}} \end{split}$$

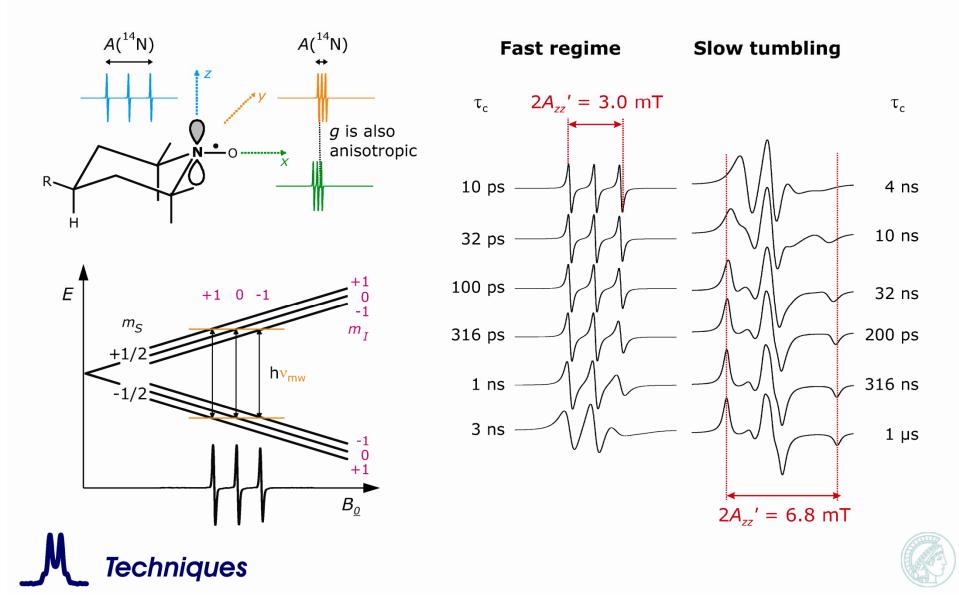
Variation in the EPR Spectrum for a Single Nitroxide



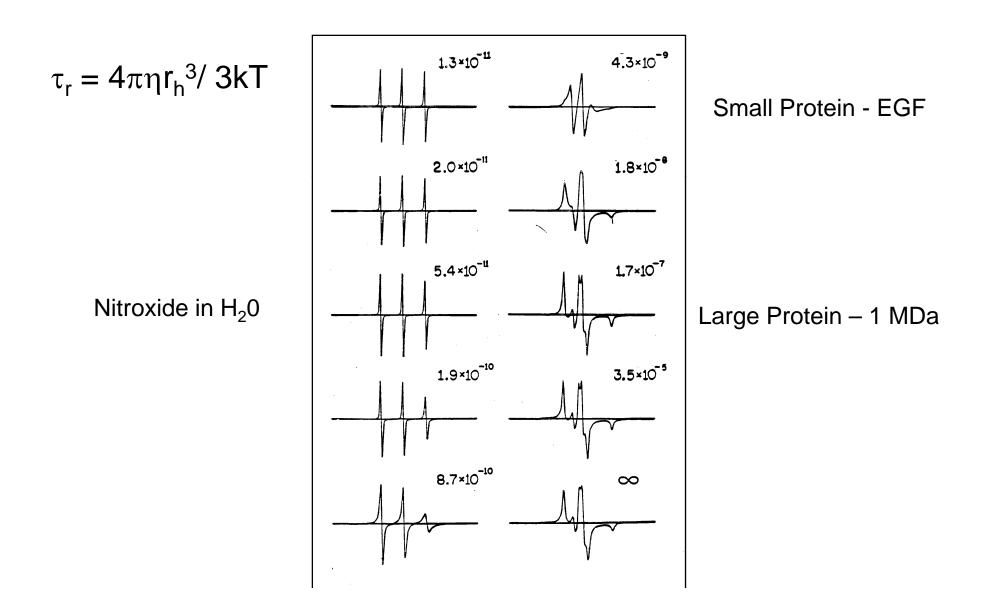
Two-Site Exchange Model Effects of Motion on EPR Spectra



Rotational correlation times



EPR Spectra as a Function of Rotational Correlation Time; $\tau_r = 1/6D_r$



Electron Spin-Spin Relaxation - Accessibility

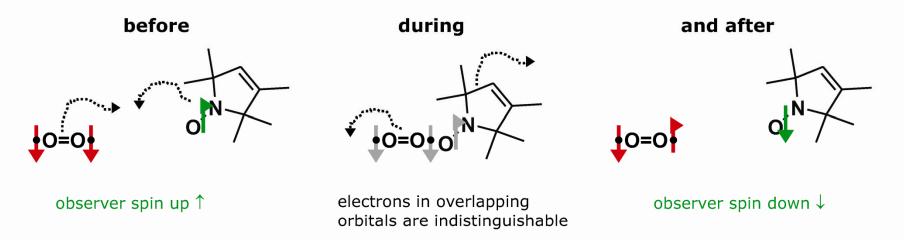
Nitroxide - slow relaxation

 T_{1e} 's of isolated Nitroxide spin labels are on the order of a few µsec's at physiological temperatures



Relaxation via Collisional Exchange

Diffusing paramagnetic species



collision

- \Rightarrow relaxation time T_1 decreases with increasing exchange rate $W_{\rm ex}$
- most easily detected via saturation curves (CW EPR)

 $\Delta P_{1/2} \propto W_{ex}/T_{2e}$

ALTENBACH et al., *Proc. Natl. Acad. Sci. USA* **91**, 1667-1671 (1994)



